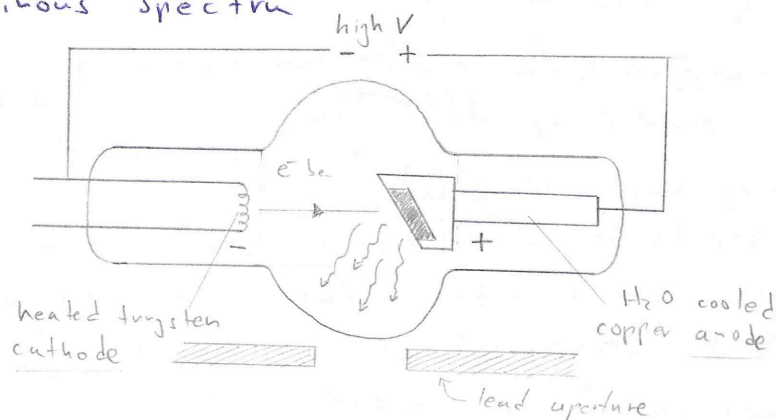
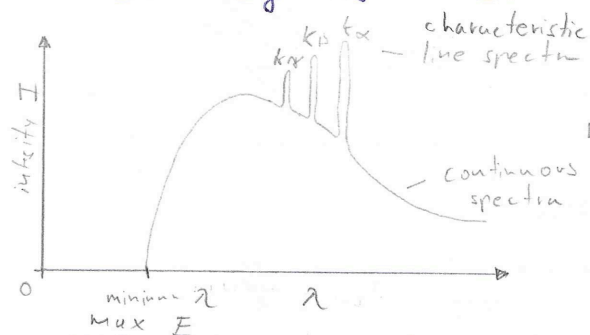


X rays

- produced when e^- are accelerated and collide with a heavy metal target. The KE of the e^- is converted into the E of an X ray photon
- breaking radiation (Bremsstrahlung) are x rays produced by the collisions. since E is conserved there is a range of frequencies which the photon can have giving a continuous spectrum

$$KE = eV \quad E_\gamma = hf \quad eV = hf$$



- in addition also certain linear spectra are produced these are characteristic to the element of the metal target
- the line spectra are produced when the e^- from beam penetrate into the target atom and knock out an e^- so e^- s from higher orbitals transition down to fill in the spot and emit X ray
- the lines have letters K, L, M, N corresponding to the energy level to which the e^- transition and subscript $\alpha, \beta, \gamma, \dots$ according to how many levels they fall
- the $K\alpha$ is the closest so e^- are most likely to transition from there $\therefore$ highest peak

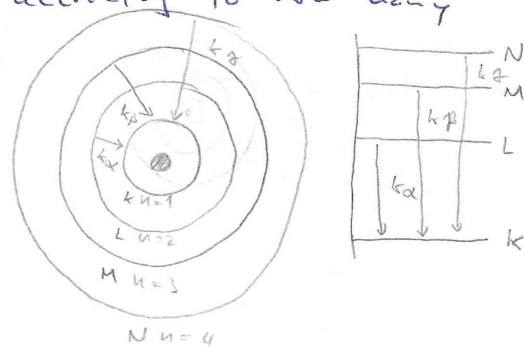
- Moseley's law

↳ discovered that f of $K\alpha$ is related to atomic number of target elements
even used it to predict and calculate Z

↳ explained by applying Bohr theory and calculating ΔE

similar to Balmer series but difficult due to effects of all e^- and shield of atom - it is $(Z-1)$ since the chge of nucleus is Ze
and chage of e^- is $-1e$ so together $(Z-1)e$

$$f = 2.5 \times 10^{15} (Z-1)^2$$



- X ray crystallography

↳ Xray λ are similar to distances between atoms in solid matter so crystals can be used as a diffraction grating producing a diffraction pattern which can be used to investigate the structure

↳ most useful is usually Bragg's reflection in which monochromatic Xrays strike crystal lattice and by Huygen-Fresnel principle lines of maxima and minima are produced which overall give a reflection pattern here the wave incident

↳ results similar to reflection but it is caused by diffraction and only occurs at specific θ

- using single crystal

↳ has to be rotated, at places where the source, atomic planes and detector align

↳ so Bragg's reflection is visible we get maxima so it is recorded spots are produced

- using powder of crystal

↳ contains particles with atomic planes at all orientations so it doesn't have to be rotated

↳ planes for which Bragg's reflection angle is ϕ create a cone of half angle ϕ so series of lines are produced

- now use the spots or lines and Bragg's formula to work out the geometry of the crystal lattice

- hard Xrays have $\downarrow \lambda, \uparrow f$ so $\uparrow E$

